

# CERTIFICATE OF ANALYSIS FOR

# Copper-Cobalt Ore (Democratic Republic of the Congo) CERTIFIED REFERENCE MATERIAL

# **OREAS 553**



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			Absolute	Standard	Deviation		Relative	Standard D	5% w	5% window		
Constituent	Certified Value		-	r	-		Ttelative			070 W		
		1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High	
Borate Fusior	1											
Al <sub>2</sub> O <sub>3</sub> , wt.%	5.58	0.061	5.46	5.70	5.40	5.76	1.08%	2.17%	3.25%	5.30	5.86	
BaO, ppm	331	67	197	465	130	532	20.25%	40.50%	60.75%	314	347	
CaO, wt.%	7.81	0.086	7.64	7.98	7.55	8.07	1.10%	2.20%	3.30%	7.42	8.20	
Co, wt.%	0.610	0.014	0.583	0.638	0.569	0.652	2.25%	4.50%	6.75%	0.580	0.641	
Cu, wt.%	0.581	0.010	0.560	0.602	0.550	0.612	1.79%	3.59%	5.38%	0.552	0.610	
Fe <sub>2</sub> O <sub>3</sub> , wt.%	1.81	0.022	1.76	1.85	1.74	1.88	1.24%	2.48%	3.72%	1.72	1.90	
K <sub>2</sub> O, wt.%	0.811	0.013	0.786	0.836	0.773	0.849	1.56%	3.11%	4.67%	0.770	0.851	
MgO, wt.%	9.20	0.144	8.91	9.49	8.77	9.63	1.57%	3.13%	4.70%	8.74	9.66	
MnO, wt.%	0.208	0.004	0.200	0.217	0.196	0.221	2.01%	4.01%	6.02%	0.198	0.219	
P <sub>2</sub> O <sub>5</sub> , wt.%	0.103	0.006	0.091	0.115	0.085	0.121	5.93%	11.87%	17.80%	0.098	0.108	
SiO <sub>2</sub> , wt.%	57.52	0.553	56.41	58.63	55.86	59.18	0.96%	1.92%	2.89%	54.64	60.39	
SO <sub>3</sub> , wt.%	0.728	0.014	0.701	0.756	0.688	0.769	1.87%	3.74%	5.62%	0.692	0.765	
SrO, ppm	< 100	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND	
TiO <sub>2</sub> , wt.%	0.347	0.012	0.323	0.372	0.311	0.384	3.51%	7.03%	10.54%	0.330	0.365	
Thermogravin	netry	1							1			
LOI <sup>1000</sup> , wt.%	14.20	0.155	13.89	14.51	13.74	14.67	1.09%	2.19%	3.28%	13.49	14.91	
Infrared Comb	oustion											
C, wt.%	3.32	0.098	3.12	3.51	3.02	3.61	2.95%	5.90%	8.85%	3.15	3.48	
S, wt.%	0.280	0.014	0.252	0.309	0.238	0.323	5.07%	10.15%	15.22%	0.266	0.294	
Sulphuric Aci	d 5% Leach								-			
Co, wt.%	0.469	0.079	0.310	0.628	0.231	0.707	16.94%	33.87%	50.81%	0.446	0.493	
Cu, wt.%	0.440	0.024	0.392	0.488	0.368	0.512	5.45%	10.91%	16.36%	0.418	0.462	
Sulphuric Aci	d 10% Leac	h										
Co, wt.%	0.470	0.078	0.314	0.626	0.236	0.704	16.57%	33.15%	49.72%	0.446	0.493	
Cu, wt.%	0.474	0.039	0.397	0.552	0.358	0.590	8.18%	16.36%	24.53%	0.450	0.498	
Peroxide Fusi	ion ICP											
Ag, ppm	< 1	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND	
Al, wt.%	2.98	0.077	2.83	3.13	2.75	3.21	2.58%	5.16%	7.74%	2.83	3.13	
B, ppm	309	21	267	351	246	372	6.80%	13.60%	20.41%	294	325	
Ba, ppm	318	16	286	350	270	366	5.06%	10.12%	15.17%	302	334	
Bi, ppm	2.53	0.37	1.79	3.26	1.42	3.63	14.58%	29.15%	43.73%	2.40	2.65	
Ca, wt.%	5.61	0.117	5.37	5.84	5.25	5.96	2.09%	4.17%	6.26%	5.32	5.89	
Ce, ppm	54	3.5	47	61	43	64	6.58%	13.16%	19.74%	51	56	
Co, wt.%	0.614	0.018	0.578	0.650	0.560	0.668	2.94%	5.89%	8.83%	0.583	0.645	
Cr, ppm	52	13	26	78	13	91	24.95%	49.90%	74.85%	50	55	
Cs, ppm	0.73	0.073	0.59	0.88	0.51	0.95	9.92%	19.84%	29.76%	0.70	0.77	
Cu, wt.%	0.583	0.012	0.558	0.608	0.546	0.620	2.14%	4.28%	6.42%	0.554	0.612	
Dy, ppm	3.66	0.299	3.07	4.26	2.77	4.56	8.17%	16.33%	24.50%	3.48	3.85	
Er, ppm	1.97	0.22	1.53	2.41	1.31	2.63	11.16%	22.32%	33.47%	1.87	2.07	
Eu, ppm	0.85	0.16	0.54	1.17	0.39	1.32	18.26%	36.53%	54.79%	0.81	0.90	
Fe, wt.%	1.26	0.033	1.20	1.33	1.16	1.36	2.59%	5.18%	7.78%	1.20	1.33	
Ga, ppm	8.69	0.697	7.30	10.09	6.60	10.78	8.02%	16.04%	24.06%	8.26	9.13	
Gd, ppm	4.60	0.51	3.58	5.61	3.07	6.12	11.06%	22.11%	33.17%	4.37	4.83	
							= 1000 pp					

Table 1. Certified Values and Performance Gates for OREAS 553.

Note 1: intervals may appear asymmetric due to rounding.

Note 2: the number of decimal places quoted does not imply accuracy of the certified value to this level but are given to minimise rounding errors when calculating 2SD and 3SD windows.



Constituent	Certified		Absolute	Standard	Deviations	S	Relative	Standard D	eviations	5% w	indow
Constituent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High
Peroxide Fus	ion ICP cont	tinued									
Ho, ppm	0.70	0.066	0.56	0.83	0.50	0.89	9.51%	19.03%	28.54%	0.66	0.73
K, wt.%	0.710	0.021	0.668	0.752	0.647	0.773	2.94%	5.87%	8.81%	0.674	0.745
La, ppm	32.0	2.60	26.7	37.2	24.1	39.8	8.15%	16.29%	24.44%	30.4	33.6
Li, ppm	103	8	88	119	80	127	7.70%	15.40%	23.10%	98	109
Lu, ppm	0.28	0.04	0.21	0.35	0.17	0.39	12.71%	25.42%	38.12%	0.27	0.30
Mg, wt.%	5.52	0.094	5.33	5.70	5.23	5.80	1.70%	3.41%	5.11%	5.24	5.79
Mn, wt.%	0.160	0.006	0.149	0.171	0.143	0.176	3.47%	6.94%	10.40%	0.152	0.168
Nb, ppm	9.25	1.82	5.60	12.89	3.78	14.72	19.70%	39.41%	59.11%	8.79	9.71
Nd, ppm	24.3	1.82	20.7	27.9	18.8	29.7	7.48%	14.95%	22.43%	23.1	25.5
P, wt.%	0.045	0.008	0.029	0.062	0.020	0.071	18.48%	36.97%	55.45%	0.043	0.048
Pb, ppm	13.6	2.8	8.0	19.1	5.3	21.9	20.40%	40.81%	61.21%	12.9	14.2
Pr, ppm	6.55	0.581	5.39	7.71	4.81	8.29	8.87%	17.75%	26.62%	6.22	6.88
Rb, ppm	25.9	1.61	22.7	29.2	21.1	30.8	6.22%	12.44%	18.66%	24.6	27.2
S, wt.%	0.283	0.020	0.243	0.323	0.223	0.343	7.04%	14.07%	21.11%	0.269	0.297
Sc, ppm	6.38	1.07	4.24	8.52	3.17	9.59	16.76%	33.52%	50.27%	6.06	6.70
Si, wt.%	27.31	0.634	26.05	28.58	25.41	29.22	2.32%	4.64%	6.96%	25.95	28.68
Sm, ppm	4.41	0.46	3.48	5.34	3.02	5.81	10.52%	21.05%	31.57%	4.19	4.63
Sr, ppm	59	3.8	52	67	48	71	6.37%	12.74%	19.11%	56	62
Ta, ppm	0.66	0.17	0.32	0.99	0.16	1.16	25.45%	50.90%	76.36%	0.63	0.69
Tb, ppm	0.64	0.08	0.48	0.79	0.40	0.87	12.43%	24.87%	37.30%	0.60	0.67
Th, ppm	5.75	0.70	4.35	7.15	3.64	7.86	12.21%	24.43%	36.64%	5.46	6.04
Ti, wt.%	0.209	0.012	0.186	0.232	0.174	0.244	5.55%	11.10%	16.64%	0.199	0.219
TI, ppm	< 0.5	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND
Tm, ppm	0.27	0.05	0.17	0.36	0.13	0.40	17.24%	34.49%	51.73%	0.25	0.28
U, ppm	27.1	2.67	21.8	32.5	19.1	35.1	9.85%	19.70%	29.56%	25.8	28.5
V, ppm	73	5.8	62	85	56	91	7.98%	15.95%	23.93%	70	77
Y, ppm	19.2	1.67	15.9	22.5	14.2	24.2	8.67%	17.35%	26.02%	18.2	20.2
Yb, ppm	1.85	0.20	1.45	2.26	1.25	2.46	10.92%	21.84%	32.76%	1.76	1.95
Zn, ppm	54	10	34	75	24	85	18.58%	37.15%	55.73%	52	57
Zr, ppm	106	14	79	134	65	147	12.88%	25.76%	38.63%	101	112
*4-Acid Diges	tion										
Ag, ppm	0.142	0.021	0.101	0.184	0.080	0.205	14.62%	29.24%	43.85%	0.135	0.149
AI, wt.%	2.93	0.077	2.78	3.09	2.70	3.17	2.64%	5.27%	7.91%	2.79	3.08
As, ppm	7.41	1.02	5.37	9.45	4.35	10.47	13.75%	27.49%	41.24%	7.04	7.78
Ba, ppm	319	12	295	343	283	355	3.74%	7.47%	11.21%	303	335
Be, ppm	1.48	0.076	1.33	1.63	1.25	1.71	5.15%	10.31%	15.46%	1.41	1.55
Bi, ppm	2.39	0.172	2.04	2.73	1.87	2.90	7.20%	14.40%	21.60%	2.27	2.51
Ca, wt.%	5.57	0.129	5.31	5.83	5.18	5.96	2.32%	4.64%	6.96%	5.29	5.85
Cd, ppm	0.19	0.014	0.16	0.22	0.15	0.24	7.39%	14.78%	22.17%	0.18	0.20
Ce, ppm	45.5	5.2	35.1	55.8	29.9	61.0	11.39%	22.79%	34.18%	43.2	47.7
Co, wt.%	0.610	0.027	0.557	0.664	0.530	0.691	4.37%	8.74%	13.12%	0.580	0.641
Cr, ppm	24.6	6.3	12.0	37.2	5.7	43.5	25.59%	51.17%	76.76%	23.4	25.8

SI unit equivalents: ppm, parts per million  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb, parts per billion.

\*Four acid digestion quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.

Note: intervals may appear asymmetric due to rounding; the number of decimal places quoted does not imply accuracy of the certified value to this level but are given to minimise rounding errors when calculating 2SD and 3SD windows.



-	Certified		Absolute	Standard	Deviations	6	Relative	Standard D	eviations	5% w	indow
Constituent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High
*4-Acid Diges	tion continu	ied									
Cs, ppm	0.69	0.061	0.56	0.81	0.50	0.87	8.86%	17.72%	26.58%	0.65	0.72
Cu, wt.%	0.582	0.016	0.550	0.613	0.534	0.629	2.71%	5.42%	8.13%	0.552	0.611
Dy, ppm	3.38	0.132	3.11	3.64	2.98	3.78	3.92%	7.84%	11.76%	3.21	3.55
Er, ppm	1.86	0.079	1.71	2.02	1.63	2.10	4.24%	8.47%	12.71%	1.77	1.96
Eu, ppm	0.82	0.031	0.76	0.88	0.73	0.91	3.73%	7.45%	11.18%	0.78	0.86
Fe, wt.%	1.24	0.040	1.16	1.32	1.12	1.36	3.25%	6.49%	9.74%	1.18	1.30
Ga, ppm	8.01	0.451	7.11	8.91	6.66	9.36	5.63%	11.25%	16.88%	7.61	8.41
Gd, ppm	4.11	0.252	3.61	4.62	3.36	4.87	6.13%	12.26%	18.38%	3.91	4.32
Hf, ppm	2.29	0.194	1.91	2.68	1.71	2.88	8.46%	16.91%	25.37%	2.18	2.41
Ho, ppm	0.66	0.054	0.55	0.77	0.50	0.82	8.17%	16.34%	24.51%	0.63	0.69
In, ppm	0.21	0.012	0.19	0.24	0.18	0.25	5.59%	11.17%	16.76%	0.20	0.22
K, wt.%	0.695	0.028	0.640	0.751	0.612	0.778	3.98%	7.97%	11.95%	0.661	0.730
La, ppm	24.9	4.0	16.9	32.9	12.9	37.0	16.08%	32.16%	48.24%	23.7	26.2
Li, ppm	99	5.0	89	109	84	114	5.01%	10.03%	15.04%	94	104
Lu, ppm	0.25	0.019	0.21	0.28	0.19	0.30	7.78%	15.57%	23.35%	0.23	0.26
Mg, wt.%	5.51	0.114	5.28	5.73	5.17	5.85	2.07%	4.13%	6.20%	5.23	5.78
Mn, wt.%	0.156	0.007	0.143	0.169	0.137	0.176	4.17%	8.34%	12.51%	0.148	0.164
Mo, ppm	2.56	0.183	2.19	2.93	2.01	3.11	7.16%	14.33%	21.49%	2.43	2.69
Na, wt.%	0.040	0.005	0.030	0.050	0.025	0.055	12.59%	25.17%	37.76%	0.038	0.042
Nb, ppm	6.17	1.10	3.97	8.37	2.87	9.47	17.83%	35.66%	53.49%	5.86	6.48
Nd, ppm	21.1	2.8	15.5	26.7	12.7	29.4	13.21%	26.43%	39.64%	20.0	22.1
Ni, ppm	25.6	1.12	23.3	27.8	22.2	28.9	4.39%	8.78%	13.17%	24.3	26.8
P, wt.%	0.045	0.002	0.040	0.049	0.037	0.052	5.51%	11.02%	16.54%	0.042	0.047
Pb, ppm	12.8	1.6	9.6	16.0	8.0	17.6	12.48%	24.96%	37.44%	12.2	13.5
Pr, ppm	5.30	0.69	3.91	6.68	3.22	7.37	13.05%	26.09%	39.14%	5.03	5.56
Rb, ppm	25.4	1.40	22.6	28.2	21.2	29.6	5.51%	11.02%	16.52%	24.1	26.6
Re, ppm	0.009	0.002	0.006	0.013	0.004	0.015	20.12%	40.25%	60.37%	0.009	0.010
S, wt.%	0.294	0.015	0.263	0.325	0.248	0.340	5.22%	10.44%	15.65%	0.279	0.309
Sb, ppm	0.41	0.035	0.34	0.48	0.30	0.51	8.52%	17.04%	25.56%	0.39	0.43
Sc, ppm	6.80	0.223	6.36	7.25	6.13	7.47	3.28%	6.56%	9.83%	6.46	7.14
Sm, ppm	4.24	0.327	3.59	4.90	3.26	5.22	7.71%	15.42%	23.13%	4.03	4.46
Sn, ppm	1.20	0.12	0.96	1.44	0.84	1.56	10.00%	20.01%	30.01%	1.14	1.26
Sr, ppm	56	2.1	52	60	50	63	3.75%	7.50%	11.24%	53	59
Ta, ppm	0.45	0.15	0.16	0.74	0.01	0.88	32.59%	65.18%	97.77%	0.42	0.47
Tb, ppm	0.60	0.049	0.50	0.70	0.45	0.74	8.18%	16.36%	24.53%	0.57	0.63
Th, ppm	5.46	0.435	4.59	6.33	4.15	6.77	7.97%	15.95%	23.92%	5.19	5.73
TI, ppm	0.12	0.012	0.10	0.15	0.09	0.16	9.57%	19.14%	28.71%	0.12	0.13
Tm, ppm	0.26	0.024	0.22	0.31	0.19	0.33	8.96%	17.92%	26.89%	0.25	0.28
U, ppm	25.5	1.85	21.8	29.2	20.0	31.1	7.24%	14.48%	21.72%	24.3	26.8
V, ppm	70	3.2	63	76	60	80	4.60%	9.19%	13.79%	66	73
W, ppm	0.87	0.15	0.57	1.17	0.42	1.32	17.18%	34.36%	51.54%	0.82	0.91
Y, ppm	17.6	1.15	15.3	19.9	14.1	21.0	6.55%	13.11%	19.66%	16.7	18.4
Y, ppm	17.6	1.15	15.3	19.9	14.1	21.0	6.55%	13.11%	19.66%	16.7	18.4

SI unit equivalents: ppm, parts per million  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb, parts per billion.

\*<u>Four acid digestion</u> quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.

Note: intervals may appear asymmetric due to rounding; the number of decimal places quoted does not imply accuracy of the certified value to this level but are given to minimise rounding errors when calculating 2SD and 3SD windows.



	Certified		Absolute	Standard	Deviation	S	Relative	Standard D	eviations	5% w	indow
Constituent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High
*4-Acid Diges	tion continu	ied									
Yb, ppm	1.68	0.085	1.51	1.85	1.42	1.93	5.09%	10.18%	15.27%	1.59	1.76
Zn, ppm	47.8	2.05	43.7	51.9	41.6	53.9	4.30%	8.59%	12.89%	45.4	50.2
Zr, ppm	80	4.7	71	90	66	94	5.81%	11.61%	17.42%	76	84
Aqua Regia D	igestion										
Ag, ppm	0.093	0.017	0.059	0.127	0.042	0.144	18.24%	36.47%	54.71%	0.088	0.098
Al, wt.%	1.43	0.114	1.21	1.66	1.09	1.77	7.94%	15.88%	23.82%	1.36	1.51
As, ppm	6.37	1.03	4.31	8.44	3.27	9.47	16.21%	32.42%	48.64%	6.06	6.69
B, ppm	10.4	0.53	9.4	11.5	8.8	12.0	5.07%	10.14%	15.20%	9.9	11.0
Ba, ppm	266	19	228	303	210	322	7.04%	14.07%	21.11%	252	279
Be, ppm	1.03	0.061	0.90	1.15	0.84	1.21	5.97%	11.93%	17.90%	0.98	1.08
Bi, ppm	1.92	0.106	1.71	2.14	1.61	2.24	5.49%	10.97%	16.46%	1.83	2.02
Ca, wt.%	5.37	0.152	5.07	5.67	4.91	5.83	2.83%	5.67%	8.50%	5.10	5.64
Cd, ppm	0.19	0.016	0.16	0.22	0.14	0.24	8.67%	17.35%	26.02%	0.18	0.20
Ce, ppm	15.6	1.6	12.4	18.8	10.8	20.4	10.32%	20.65%	30.97%	14.8	16.4
Co, wt.%	0.585	0.022	0.541	0.628	0.520	0.649	3.70%	7.39%	11.09%	0.555	0.614
Cr, ppm	21.4	1.59	18.2	24.6	16.7	26.2	7.40%	14.81%	22.21%	20.3	22.5
Cs, ppm	0.14	0.04	0.06	0.21	0.03	0.25	26.91%	53.83%	80.74%	0.13	0.14
Cu, wt.%	0.575	0.014	0.548	0.603	0.534	0.616	2.39%	4.78%	7.16%	0.546	0.604
Dy, ppm	1.91	0.084	1.75	2.08	1.66	2.16	4.36%	8.73%	13.09%	1.82	2.01
Er, ppm	0.84	0.049	0.74	0.94	0.70	0.99	5.81%	11.62%	17.43%	0.80	0.88
Eu, ppm	0.51	0.032	0.45	0.58	0.42	0.61	6.13%	12.25%	18.38%	0.49	0.54
Fe, wt.%	1.03	0.038	0.95	1.10	0.92	1.14	3.66%	7.32%	10.99%	0.98	1.08
Ga, ppm	4.22	0.384	3.45	4.99	3.07	5.37	9.11%	18.22%	27.33%	4.01	4.43
Gd, ppm	2.62	0.29	2.04	3.20	1.75	3.49	11.10%	22.21%	33.31%	2.49	2.75
Ge, ppm	< 0.1	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND
Hf, ppm	0.30	0.05	0.20	0.40	0.15	0.45	16.29%	32.57%	48.86%	0.29	0.32
Hg, ppm	0.036	0.008	0.021	0.052	0.013	0.060	21.81%	43.62%	65.43%	0.035	0.038
Ho, ppm	0.33	0.028	0.27	0.38	0.24	0.41	8.48%	16.97%	25.45%	0.31	0.34
In, ppm	0.18	0.008	0.17	0.20	0.16	0.21	4.33%	8.66%	12.99%	0.18	0.19
K, wt.%	0.129	0.012	0.104	0.153	0.092	0.165	9.51%	19.02%	28.53%	0.122	0.135
La, ppm	5.35	0.96	3.42	7.27	2.46	8.24	18.01%	36.01%	54.02%	5.08	5.62
Li, ppm	79	9	62	96	53	105	11.01%	22.02%	33.03%	75	83
Lu, ppm	0.096	0.005	0.087	0.106	0.082	0.110	4.85%	9.70%	14.55%	0.091	0.101
Mg, wt.%	5.15	0.292	4.57	5.74	4.28	6.03	5.67%	11.35%	17.02%	4.89	5.41
Mn, wt.%	0.153	0.008	0.137	0.170	0.128	0.178	5.44%	10.89%	16.33%	0.146	0.161
Mo, ppm	2.41	0.175	2.06	2.75	1.88	2.93	7.26%	14.52%	21.78%	2.29	2.53
Nb, ppm	< 0.05	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND
Nd, ppm	7.64	1.11	5.42	9.86	4.31	10.97	14.54%	29.09%	43.63%	7.26	8.02
Ni, ppm	23.4	1.72	20.0	26.9	18.3	28.6	7.32%	14.64%	21.97%	22.3	24.6
P, wt.%	0.040	0.003	0.033	0.047	0.030	0.051	8.63%	17.25%	25.88%	0.038	0.042
Pb, ppm	11.3	0.81	9.7	13.0	8.9	13.8	7.17%	14.35%	21.52%	10.8	11.9
Pr, ppm	1.60	0.27	1.07	2.14	0.80	2.41	16.74%	33.49%	50.23%	1.52	1.68
			1				1				

SI unit equivalents: ppm, parts per million  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb, parts per billion.

\*<u>Four acid digestion</u> quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.

Note: intervals may appear asymmetric due to rounding; the number of decimal places quoted does not imply accuracy of the certified value to this level but are given to minimise rounding errors when calculating 2SD and 3SD windows.



Constituent	Certified		Absolute	Standard	Deviations	6	Relative	Standard D	eviations	5% w	indow
Constituent	Value	1SD	2SD Low	2SD High	3SD Low	3SD High	1RSD	2RSD	3RSD	Low	High
Aqua Regia D	igestion co	ntinued									
Rb, ppm	4.21	0.413	3.38	5.03	2.97	5.45	9.82%	19.63%	29.45%	4.00	4.42
Re, ppm	0.009	0.001	0.007	0.011	0.006	0.012	12.22%	24.43%	36.65%	0.008	0.009
S, wt.%	0.270	0.015	0.240	0.300	0.225	0.315	5.56%	11.12%	16.69%	0.256	0.283
Sb, ppm	0.23	0.03	0.18	0.29	0.15	0.32	11.95%	23.91%	35.86%	0.22	0.25
Sc, ppm	4.79	0.52	3.76	5.82	3.25	6.34	10.75%	21.50%	32.25%	4.55	5.03
Se, ppm	1.40	0.31	0.77	2.02	0.46	2.34	22.36%	44.72%	67.08%	1.33	1.47
Sm, ppm	2.37	0.190	1.99	2.75	1.80	2.95	8.01%	16.03%	24.04%	2.26	2.49
Sn, ppm	0.36	0.06	0.23	0.49	0.17	0.55	17.55%	35.11%	52.66%	0.34	0.38
Sr, ppm	41.2	3.08	35.0	47.3	32.0	50.4	7.47%	14.95%	22.42%	39.1	43.3
Ta, ppm	< 0.01	IND	IND	IND	IND	IND	IND	IND	IND	IND	IND
Tb, ppm	0.38	0.019	0.34	0.41	0.32	0.43	4.94%	9.89%	14.83%	0.36	0.40
Th, ppm	2.21	0.181	1.85	2.57	1.67	2.75	8.21%	16.41%	24.62%	2.10	2.32
TI, ppm	0.063	0.007	0.050	0.076	0.043	0.083	10.48%	20.97%	31.45%	0.060	0.066
U, ppm	22.9	1.50	19.9	25.9	18.4	27.4	6.55%	13.09%	19.64%	21.7	24.0
V, ppm	38.8	2.28	34.3	43.4	32.0	45.6	5.86%	11.72%	17.59%	36.9	40.8
W, ppm	0.35	0.04	0.27	0.43	0.22	0.48	12.07%	24.15%	36.22%	0.33	0.37
Y, ppm	9.15	0.352	8.44	9.85	8.09	10.20	3.85%	7.69%	11.54%	8.69	9.60
Yb, ppm	0.70	0.055	0.59	0.81	0.53	0.87	7.90%	15.81%	23.71%	0.66	0.73
Zn, ppm	44.4	2.84	38.7	50.1	35.9	52.9	6.40%	12.80%	19.20%	42.2	46.6
Zr, ppm	10.3	0.75	8.9	11.8	8.1	12.6	7.23%	14.47%	21.70%	9.8	10.9

SI unit equivalents: ppm, parts per million  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb, parts per billion.

Note: intervals may appear asymmetric due to rounding; the number of decimal places quoted does not imply accuracy of the certified value to this level but are given to minimise rounding errors when calculating 2SD and 3SD windows.

#### Table 2. Indicative Values for OREAS 553.

Constituent	Unit	Value	Constituent	Unit	Value	Constituent	Unit	Value
Borate Fusio	on XRF							
As	ppm	21.7	Мо	ppm	30.5	Sn	ppm	< 10
Bi	ppm	< 10	Na <sub>2</sub> O	wt.%	0.045	V <sub>2</sub> O <sub>5</sub>	ppm	143
Cd	ppm	< 10	Ni	ppm	87	Zn	ppm	56
CI	ppm	265	Pb	ppm	113	Zr	ppm	85
Cr <sub>2</sub> O <sub>3</sub>	ppm	65	Sb	ppm	< 10			
Thermograv	imetry							
H <sub>2</sub> O-	wt.%	0.245						
Peroxide Fu	sion ICP							
As	ppm	9.80	Hg	ppm	< 0.1	Sb	ppm	1.15
Be	ppm	1.85	In	ppm	0.22	Se	ppm	6.72
Cd	ppm	< 10	Мо	ppm	3.62	Sn	ppm	1.33
Ge	ppm	0.94	Ni	ppm	36.9	Те	ppm	< 1
Hf	ppm	3.08	Re	ppm	< 0.1	W	ppm	1.34
4-Acid Diges	stion							
В	ppm	10.0	Hg	ppm	0.045	Те	ppm	0.038
Ge	ppm	0.095	Se	ppm	1.64	Ti	wt.%	0.127

SI unit equivalents: ppm (parts per million)  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb (parts per billion).

Note: the number of significant figures reported is not a reflection of the level of certainty of stated values. They are instead an artefact of ORE's in-house CRM-specific LIMS.



	Table 2 continued.											
Constituent	Unit	Value	Constituent	Unit	Value	Constituent	Unit	Value				
Aqua Regia Digestion												
Au	ppm	0.008	Pt	ppb	13.3	Ti	wt.%	0.003				
Na	wt.%	0.017	Si	wt.%	0.097	Tm	ppm	0.11				
Pd	ppb	< 10	Те	ppm	0.031							

Note: the number of significant figures reported is not a reflection of the level of certainty of stated values. They are instead an artefact of ORE's in-house CRM-specific LIMS.

#### INTRODUCTION

OREAS reference materials are intended to provide a low-cost method of evaluating and improving the quality of analysis of geological samples. To the geologist they provide a means of implementing quality control in analytical data sets generated in exploration from the grass roots level through to prospect evaluation, and in grade control at mining operations. To the analyst they provide an effective means of calibrating analytical equipment, assessing new techniques and routinely monitoring in-house procedures. OREAS reference materials enable users to successfully achieve process control of these tasks because the observed variance from repeated analysis has its origin almost exclusively in the analytical process rather than the reference material itself. In evaluating laboratory performance with this CRM, the section headed 'Instructions for correct use' should be read carefully.

Table 1 provides performance gate intervals for the certified values. Table 2 shows indicative values, Table 3 provides some indicative physical properties and Table 4 presents the 95% confidence and tolerance limits for all certified values. Tabulated results of all elements together with uncorrected means, medians, standard deviations, relative standard deviations and per cent deviation of lab means from the corrected mean of means (PDM<sup>3</sup>) are presented in the detailed certification data for this CRM (**OREAS 553-DataPack.1.1.210420\_171018.xlsx**).

Results are also presented in scatter plots for Co and Cu by 4-acid digestion and peroxide fusion ICP in Figures 1 to 4 respectively, together with  $\pm 3$ SD (magenta) and  $\pm 5\%$  (yellow) control lines and certified value (green line). Accepted individual results are coloured blue and individual and dataset outliers are identified in red and violet, respectively.

#### SOURCE MATERIAL

OREAS 553 was prepared from a blend of copper-cobalt oxide ore materials supplied by Metorex (subsidiary of Jinchuan Group) from the Dilala Est Copper Cobalt Project, located within the Kolwezi Klippe area of Lualaba Province (Democratic Republic of the Congo). Kolwezi is ~320 kilometres northwest from Lubumbashi, the provincial capital of Haut Katanga Province. Copper-cobalt ores within the Kolwezi Klippe area are stratiform, sediment-hosted deposits hosted within Neoproterozoic sedimentary rocks of the Central African Copperbelt. Primary mineralisation typically consists of bornite, chalcopyrite and cobalt sulphides hosted within dolomitic argillite, shales, sandstone and dolomite horizons. Significant oxide minerals include malachite, chalcocite, digenite and hetrogenite.



## PERFORMANCE GATES

Table 1 above shows intervals calculated for two and three standard deviations. As a guide these intervals may be regarded as warning or rejection for multiple 2SD outliers, or rejection for individual 3SD outliers in QC monitoring, although their precise application should be at the discretion of the QC manager concerned (also see 'Intended Use' section below). Westgard Rules extend the basics of single-rule QC monitoring using multi-rules (for more information visit www.westgard.com/mltirule.htm). A second method utilises a 5% window calculated directly from the certified value.

Standard deviation is also shown in relative percent for one, two and three relative standard deviations (1RSD, 2RSD and 3RSD) to facilitate an appreciation of the magnitude of these numbers and a comparison with the 5% window. Caution should be exercised when concentration levels approach lower limits of detection of the analytical methods employed as performance gates calculated from standard deviations tend to be excessively wide whereas those determined by the 5% method are too narrow. One approach used at commercial laboratories is to set the acceptance criteria at twice the detection level (DL)  $\pm$  10%.

*i.e.* Certified Value ± 10% ± 2DL (adapted from Govett, 1983).

## **COMMINUTION AND HOMOGENISATION PROCEDURES**

The material constituting OREAS 553 was prepared in the following manner:

- Drying to constant mass at 105° Celsius;
- Multi-stage milling of each source material to achieve a particle size distribution of >98% passing 75 microns;
- Preliminary homogenisation of each source material;
- Representative sampling and check assaying of each source material;
- Blending in appropriate proportions to achieve target grades;
- Packaging in 10g units sealed in laminated foil pouches and 1000g units in plastic jars.

## PHYSICAL PROPERTIES

OREAS 553 was tested at ORE Research & Exploration Pty Ltd's onsite facility for various physical properties. Table 3 presents these findings that should be used for informational purposes only.

Bulk Density (g/L)	Moisture%	Munsell Notation <sup>‡</sup>	Munsell Color‡
845.4	0.36	5YR 8/1	Pinkish Gray

#### Table 3. Physical properties of OREAS 553.

<sup>‡</sup>The Munsell Rock Color Chart helps geologists and archeologists communicate with colour more effectively by cross-referencing ISCC-NBS colour names with unique Munsell alpha-numeric colour notations for rock colour samples.



# ANALYTICAL PROGRAM

Thirty analytical laboratories participated in the program to characterise the elements reported in Table 1. The following methods were employed:

- Lithium borate fusion whole rock analysis package by X-ray fluorescence (up to 19 laboratories depending on the element);
- Thermogravimetry: Loss on Ignition (LOI) at 1000°C (13 laboratories used a thermogravimetric analyser, 7 laboratories included LOI with their fusion package and 4 laboratories used conventional muffle furnace);
- C and S by infrared combustion furnace/CS analyser (27 laboratories);
- Cu and Co by \*5% sulphuric acid leach with ICP or AAS finish (up to 19 laboratories);
- Cu and Co by \*10% sulphuric acid leach with ICP or AAS finish (up to 12 laboratories);
- Sodium peroxide fusion with full suite elemental package by ICP-OES and/or MS finish (up to 20 laboratories depending on the element);
- 4-acid digestion for full suite elemental package by ICP-OES and MS finish (up to 26 laboratories depending on the element);
- Aqua regia digestion for full elemental suite ICP-OES and ICP-MS (up to 26 laboratories depending on the element).

\*See 'Appendix' for specified methodology

For the round robin program ten 400g test units were taken at predetermined intervals immediately following homogenisation and are considered representative of the entire prepared batch. The six samples received by each laboratory were obtained by taking two 25g scoop splits from each of three separate 400g test units. This format enabled nested ANOVA treatment of the results to evaluate homogeneity, i.e., to ascertain whether between-unit variance is greater than within-unit variance (see 'Homogeneity Evaluation' section below).

## STATISTICAL ANALYSIS

**Standard Deviation** intervals (see Table 1) provide an indication of a level of performance that might reasonably be expected from a laboratory being monitored by this CRM in a QA/QC program. They take into account errors attributable to measurement uncertainty and CRM variability. For an effective CRM the contribution of the latter should be negligible in comparison to measurement errors. The Standard Deviation values include all sources of measurement uncertainty: between-laboratory variance, within-run variance (precision errors) and CRM variability.

In the application of SD's in monitoring performance it is important to note that not all laboratories function at the same level of proficiency and that different methods in use at a particular laboratory have differing levels of precision. Each laboratory has its own inherent SD (for a specific concentration level and analyte-method pair) based on the analytical process and this SD is not directly related to the round robin program (see Intended Use section for more detail).

The SD for each analyte's certified value is calculated from the same filtered data set used to determine the certified value, i.e., after removal of all individual, lab dataset (batch) and



3SD outliers (single iteration). These outliers can only be removed after the absolute homogeneity of the CRM has been independently established, i.e., the outliers must be confidently deemed to be analytical rather than arising from inhomogeneity of the CRM. *The standard deviation is then calculated for each analyte from the pooled accepted analyses generated from the certification program.* 

**Certified Values, Confidence Limits and Tolerance Limits** (Table 4) have been determined for each analyte following removal of individual, laboratory dataset (batch) and 3SD outliers (single iteration).

**95% Confidence Limits** are inversely proportional to the number of participating laboratories and inter-laboratory agreement. It is a measure of the reliability of the certified value. A 95% confidence interval indicates a 95% probability that the true value of the analyte under consideration lies between the upper and lower limits. **95% Confidence Limits should not be used as control limits for laboratory performance.** 

**Indicative (uncertified) values** (Table 2) are provided where i) a laboratory reported analytes beyond those requested; ii) the number of laboratories reporting a particular analyte is insufficient (< 5) to support certification; iii) inter-laboratory consensus is poor; or iv) a significant proportion of results are outlying.

For individual outliers within a laboratory batch the z-score test is used in combination with a second method that determines the per cent deviation of the individual value from the batch median. Outliers in general are selected on the basis of z-scores > 2.5 and with per cent deviations (i) > 3 and (ii) more than three times the average absolute per cent deviation for the batch. In certain instances, statistician's prerogative has been employed in discriminating outliers. Each laboratory data set mean is tested for outlying status based on z-score discrimination and rejected if > 2.5. After individual and laboratory data set (batch) outliers have been eliminated a non-iterative 3 standard deviation filter is applied, with those values lying outside this window also relegated to outlying status.

Constituent	Certified	95% Confid	lence Limits	95% Toler	ance Limits				
Constituent	Value	Low	High	Low	High				
Borate Fusion XRF									
Al <sub>2</sub> O <sub>3</sub> , Aluminium(III) oxide (wt.%)	5.58	5.55	5.61	5.54	5.62				
BaO, Barium oxide (ppm)	331	290	372	302	360				
CaO, Calcium oxide (wt.%)	7.81	7.77	7.85	7.76	7.86				
Co, Cobalt (wt.%)	0.610	0.603	0.618	0.602	0.619				
Cu, Copper (wt.%)	0.581	0.575	0.587	0.566	0.596				
Fe <sub>2</sub> O <sub>3</sub> , Iron(III) oxide (wt.%)	1.81	1.80	1.82	1.79	1.83				
K <sub>2</sub> O, Potassium oxide (wt.%)	0.811	0.806	0.816	0.799	0.823				
MgO, Magnesium oxide (wt.%)	9.20	9.12	9.27	9.14	9.25				
MnO, Manganese oxide (wt.%)	0.208	0.207	0.210	0.205	0.212				
P <sub>2</sub> O <sub>5</sub> , Phosphorus(V) oxide (wt.%)	0.103	0.099	0.107	0.101	0.105				
SiO <sub>2</sub> , Silicon dioxide (wt.%)	57.52	57.28	57.76	57.31	57.73				
SO <sub>3</sub> , Sulphur trioxide (wt.%)	0.728	0.721	0.736	0.711	0.746				

 Table 4. 95% Confidence & Tolerance Limits for OREAS 553.

SI unit equivalents: ppm, parts per million  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb, parts per billion. Note: intervals may appear asymmetric due to rounding.



Constituent	Certified	95% Confid	ence Limits	95% Tolerance Limits	
Constituent	Value	Low	High	Low	High
Thermogravimetry					
LOI <sup>1000</sup> , Loss On Ignition @1000°C (wt.%)	14.20	14.13	14.27	14.15	14.25
Infrared Combustion			1		1
C, Carbon (wt.%)	3.32	3.28	3.35	3.29	3.34
S, Sulphur (wt.%)	0.280	0.275	0.286	0.269	0.292
Sulphuric Acid 5% Leach			1		1
Co, Cobalt (wt.%)	0.469	0.406	0.533	0.457	0.481
Cu, Copper (wt.%)	0.440	0.428	0.452	0.433	0.447
Sulphuric Acid 10% Leach			1		I
Co, Cobalt (wt.%)	0.470	0.422	0.518	0.451	0.489
Cu, Copper (wt.%)	0.474	0.447	0.501	0.463	0.485
Peroxide Fusion ICP			1		
Ag, Silver (ppm)	< 1	IND	IND	IND	IND
Al, Aluminium (wt.%)	2.98	2.95	3.01	2.91	3.05
B, Boron (ppm)	309	290	328	289	329
Ba, Barium (ppm)	318	310	326	306	330
Bi, Bismuth (ppm)	2.53	2.22	2.84	2.38	2.68
Ca, Calcium (wt.%)	5.61	5.56	5.65	5.49	5.72
Ce, Cerium (ppm)	54	51	56	51	56
Co, Cobalt (wt.%)	0.614	0.606	0.622	0.603	0.625
Cr, Chromium (ppm)	52	45	59	47	57
Cs, Caesium (ppm)	0.73	0.69	0.77	IND	IND
Cu, Copper (wt.%)	0.583	0.578	0.587	0.571	0.595
Dy, Dysprosium (ppm)	3.66	3.49	3.84	3.44	3.89
Er, Erbium (ppm)	1.97	1.82	2.11	1.80	2.14
Eu, Europium (ppm)	0.85	0.77	0.94	0.72	0.98
Fe, Iron (wt.%)	1.26	1.25	1.27	1.23	1.29
Ga, Gallium (ppm)	8.69	8.34	9.04	8.17	9.21
Gd, Gadolinium (ppm)	4.60	4.19	5.00	4.24	4.96
Ho, Holmium (ppm)	0.70	0.64	0.75	0.64	0.76
K, Potassium (wt.%)	0.710	0.701	0.719	0.681	0.739
La, Lanthanum (ppm)	32.0	30.1	33.8	30.5	33.4
Li, Lithium (ppm)	103	99	108	99	108
Lu, Lutetium (ppm)	0.28	0.25	0.31	0.24	0.32
Mg, Magnesium (wt.%)	5.52	5.49	5.54	5.40	5.63
Mn, Manganese (wt.%)	0.160	0.157	0.162	0.155	0.164
Nb, Niobium (ppm)	9.25	7.88	10.62	IND	IND
Nd, Neodymium (ppm)	24.3	23.1	25.5	23.2	25.4
P, Phosphorus (wt.%)	0.045	0.038	0.053	IND	IND



	Certified	95% Confidence Limits		95% Tolerance Limits	
Constituent	Value	Low	High	Low	High
Peroxide Fusion ICP continued					
Pb, Lead (ppm)	13.6	11.2	15.9	12.0	15.1
Pr, Praseodymium (ppm)	6.55	5.94	7.16	6.14	6.96
Rb, Rubidium (ppm)	25.9	25.2	26.6	24.6	27.2
S, Sulphur (wt.%)	0.283	0.273	0.294	0.267	0.299
Sc, Scandium (ppm)	6.38	5.24	7.52	IND	IND
Si, Silicon (wt.%)	27.31	26.95	27.67	26.79	27.84
Sm, Samarium (ppm)	4.41	4.11	4.71	3.95	4.87
Sr, Strontium (ppm)	59	57	61	57	62
Ta, Tantalum (ppm)	0.66	0.49	0.82	IND	IND
Tb, Terbium (ppm)	0.64	0.59	0.68	0.56	0.71
Th, Thorium (ppm)	5.75	5.35	6.15	5.39	6.11
Ti, Titanium (wt.%)	0.209	0.204	0.214	0.202	0.216
TI, Thallium (ppm)	< 0.5	IND	IND	IND	IND
Tm, Thulium (ppm)	0.27	0.24	0.29	0.24	0.29
U, Uranium (ppm)	27.1	25.9	28.3	25.1	29.2
V, Vanadium (ppm)	73	69	77	69	78
Y, Yttrium (ppm)	19.2	18.3	20.2	18.4	20.0
Yb, Ytterbium (ppm)	1.85	1.74	1.97	1.64	2.07
Zn, Zinc (ppm)	54	49	60	51	58
Zr, Zirconium (ppm)	106	89	123	97	116
*4-Acid Digestion	·				
Ag, Silver (ppm)	0.142	0.131	0.153	IND	IND
Al, Aluminium (wt.%)	2.93	2.90	2.96	2.88	2.98
As, Arsenic (ppm)	7.41	6.75	8.07	7.09	7.73
Ba, Barium (ppm)	319	314	324	312	326
Be, Beryllium (ppm)	1.48	1.45	1.51	1.43	1.53
Bi, Bismuth (ppm)	2.39	2.31	2.46	2.31	2.47
Ca, Calcium (wt.%)	5.57	5.52	5.62	5.47	5.66
Cd, Cadmium (ppm)	0.19	0.19	0.20	0.17	0.22
Ce, Cerium (ppm)	45.5	43.1	47.9	44.3	46.6
Co, Cobalt (wt.%)	0.610	0.599	0.622	0.598	0.623
Cr, Chromium (ppm)	24.6	21.8	27.3	22.7	26.5
Cs, Caesium (ppm)	0.69	0.66	0.72	0.66	0.71
Cu, Copper (wt.%)	0.582	0.576	0.587	0.572	0.591
Dy, Dysprosium (ppm)	3.38	3.29	3.46	3.26	3.50
Er, Erbium (ppm)	1.86	1.81	1.92	1.79	1.94
Eu, Europium (ppm)	0.82	0.80	0.85	0.78	0.86

\*Four acid digestion quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.



Constituent	Certified	95% Confidence Limits		95% Tolerance Limits	
Constituent	Value	Low	High	Low	High
*4-Acid Digestion continued					
Fe, Iron (wt.%)	1.24	1.22	1.26	1.22	1.27
Ga, Gallium (ppm)	8.01	7.82	8.20	7.83	8.19
Gd, Gadolinium (ppm)	4.11	3.95	4.28	3.91	4.32
Hf, Hafnium (ppm)	2.29	2.20	2.39	2.18	2.41
Ho, Holmium (ppm)	0.66	0.61	0.71	0.64	0.68
In, Indium (ppm)	0.21	0.21	0.22	0.20	0.22
K, Potassium (wt.%)	0.695	0.683	0.708	0.684	0.706
La, Lanthanum (ppm)	24.9	23.1	26.7	24.2	25.6
Li, Lithium (ppm)	99	97	102	97	101
Lu, Lutetium (ppm)	0.25	0.23	0.26	0.23	0.26
Mg, Magnesium (wt.%)	5.51	5.47	5.55	5.40	5.62
Mn, Manganese (wt.%)	0.156	0.154	0.159	0.153	0.160
Mo, Molybdenum (ppm)	2.56	2.48	2.63	2.44	2.67
Na, Sodium (wt.%)	0.040	0.037	0.042	0.038	0.042
Nb, Niobium (ppm)	6.17	5.65	6.69	5.89	6.45
Nd, Neodymium (ppm)	21.1	19.3	22.9	20.4	21.7
Ni, Nickel (ppm)	25.6	25.1	26.0	24.7	26.5
P, Phosphorus (wt.%)	0.045	0.044	0.045	0.043	0.046
Pb, Lead (ppm)	12.8	12.0	13.7	12.1	13.6
Pr, Praseodymium (ppm)	5.30	4.82	5.77	5.11	5.48
Rb, Rubidium (ppm)	25.4	24.8	26.0	24.7	26.0
Re, Rhenium (ppm)	0.009	0.008	0.010	IND	IND
S, Sulphur (wt.%)	0.294	0.288	0.301	0.285	0.303
Sb, Antimony (ppm)	0.41	0.39	0.42	0.37	0.44
Sc, Scandium (ppm)	6.80	6.69	6.91	6.65	6.95
Sm, Samarium (ppm)	4.24	4.03	4.46	4.09	4.39
Sn, Tin (ppm)	1.20	1.15	1.25	IND	IND
Sr, Strontium (ppm)	56	55	57	55	57
Ta, Tantalum (ppm)	0.45	0.37	0.53	0.41	0.48
Tb, Terbium (ppm)	0.60	0.56	0.64	0.58	0.62
Th, Thorium (ppm)	5.46	5.28	5.64	5.31	5.61
TI, Thallium (ppm)	0.12	0.12	0.13	IND	IND
Tm, Thulium (ppm)	0.26	0.24	0.29	0.24	0.28
U, Uranium (ppm)	25.5	25.0	26.1	24.0	27.1
V, Vanadium (ppm)	70	68	71	68	72
W, Tungsten (ppm)	0.87	0.81	0.93	IND	IND
Y, Yttrium (ppm)	17.6	17.1	18.1	17.2	17.9

\*<u>Four acid digestion</u> quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.



Constituent	Certified	95% Confidence Limits		95% Tolerance Limits	
Constituent	Value	Low	High	Low	High
*4-Acid Digestion continued					
Yb, Ytterbium (ppm)	1.68	1.62	1.73	1.59	1.77
Zn, Zinc (ppm)	47.8	46.9	48.6	45.9	49.7
Zr, Zirconium (ppm)	80	78	82	77	84
Aqua Regia Digestion					
Ag, Silver (ppm)	0.093	0.085	0.101	0.074	0.112
Al, Aluminium (wt.%)	1.43	1.38	1.48	1.39	1.47
As, Arsenic (ppm)	6.37	5.74	7.01	6.01	6.74
B, Boron (ppm)	10.4	9.8	11.1	IND	IND
Ba, Barium (ppm)	266	257	275	258	274
Be, Beryllium (ppm)	1.03	1.01	1.05	0.96	1.10
Bi, Bismuth (ppm)	1.92	1.88	1.97	1.86	1.98
Ca, Calcium (wt.%)	5.37	5.31	5.43	5.28	5.46
Cd, Cadmium (ppm)	0.19	0.18	0.20	0.16	0.22
Ce, Cerium (ppm)	15.6	14.8	16.4	15.1	16.1
Co, Cobalt (wt.%)	0.585	0.574	0.595	0.571	0.598
Cr, Chromium (ppm)	21.4	20.7	22.1	20.6	22.3
Cs, Caesium (ppm)	0.14	0.12	0.16	IND	IND
Cu, Copper (wt.%)	0.575	0.570	0.581	0.568	0.583
Dy, Dysprosium (ppm)	1.91	1.82	2.00	IND	IND
Er, Erbium (ppm)	0.84	0.79	0.90	IND	IND
Eu, Europium (ppm)	0.51	0.49	0.54	IND	IND
Fe, Iron (wt.%)	1.03	1.01	1.05	1.01	1.05
Ga, Gallium (ppm)	4.22	4.02	4.42	4.04	4.40
Gd, Gadolinium (ppm)	2.62	2.23	3.00	2.45	2.78
Ge, Germanium (ppm)	< 0.1	IND	IND	IND	IND
Hf, Hafnium (ppm)	0.30	0.28	0.33	0.28	0.32
Hg, Mercury (ppm)	0.036	0.032	0.040	IND	IND
Ho, Holmium (ppm)	0.33	0.31	0.34	IND	IND
In, Indium (ppm)	0.18	0.18	0.19	0.17	0.20
K, Potassium (wt.%)	0.129	0.123	0.134	0.125	0.132
La, Lanthanum (ppm)	5.35	4.89	5.81	5.12	5.58
Li, Lithium (ppm)	79	74	83	77	81
Lu, Lutetium (ppm)	0.096	0.092	0.101	IND	IND
Mg, Magnesium (wt.%)	5.15	5.02	5.28	5.07	5.24
Mn, Manganese (wt.%)	0.153	0.150	0.157	0.151	0.156
Mo, Molybdenum (ppm)	2.41	2.33	2.48	2.31	2.50
Nb, Niobium (ppm)	< 0.05	IND	IND	IND	IND

\*<u>Four acid digestion</u> quantitatively dissolves nearly all minerals in the majority of geological samples however, some refractory minerals may only be partially digested.



Table 4 continued.							
Constituent	Certified 95% Cor		dence Limits	95% Toler	ance Limits		
Constituent	Value	Low	High	Low	High		
Aqua Regia Digestion continued							
Nd, Neodymium (ppm)	7.64	6.40	8.88	6.62	8.66		
Ni, Nickel (ppm)	23.4	22.7	24.2	22.6	24.3		
P, Phosphorus (wt.%)	0.040	0.039	0.042	0.039	0.041		
Pb, Lead (ppm)	11.3	10.9	11.7	10.8	11.9		
Pr, Praseodymium (ppm)	1.60	1.23	1.98	IND	IND		
Rb, Rubidium (ppm)	4.21	4.00	4.42	4.00	4.42		
Re, Rhenium (ppm)	0.009	0.009	0.009	IND	IND		
S, Sulphur (wt.%)	0.270	0.263	0.277	0.260	0.280		
Sb, Antimony (ppm)	0.23	0.22	0.25	0.22	0.25		
Sc, Scandium (ppm)	4.79	4.52	5.06	4.59	4.99		
Se, Selenium (ppm)	1.40	1.25	1.54	IND	IND		
Sm, Samarium (ppm)	2.37	2.14	2.61	2.17	2.57		
Sn, Tin (ppm)	0.36	0.33	0.39	0.32	0.40		
Sr, Strontium (ppm)	41.2	39.8	42.6	40.0	42.4		
Ta, Tantalum (ppm)	< 0.01	IND	IND	IND	IND		
Tb, Terbium (ppm)	0.38	0.36	0.39	0.36	0.39		
Th, Thorium (ppm)	2.21	2.13	2.29	2.10	2.32		
Tl, Thallium (ppm)	0.063	0.060	0.066	IND	IND		
U, Uranium (ppm)	22.9	22.4	23.3	21.3	24.4		
V, Vanadium (ppm)	38.8	37.8	39.9	37.6	40.0		
W, Tungsten (ppm)	0.35	0.33	0.37	0.30	0.40		
Y, Yttrium (ppm)	9.15	8.97	9.33	8.90	9.40		
Yb, Ytterbium (ppm)	0.70	0.65	0.75	IND	IND		
Zn, Zinc (ppm)	44.4	43.1	45.7	42.8	45.9		
Zr, Zirconium (ppm)	10.3	10.0	10.7	10.0	10.7		

#### 

SI unit equivalents: ppm, parts per million  $\equiv$  mg/kg  $\equiv$  µg/g  $\equiv$  0.0001 wt.%  $\equiv$  1000 ppb, parts per billion.

Note: intervals may appear asymmetric due to rounding; IND = indeterminate (due to reading resolution error).

#### Homogeneity Evaluation

The tolerance limits (ISO 16269:2014) shown in Table 4 were determined using an analysis of precision errors method and are considered a conservative estimate of true homogeneity. The meaning of tolerance limits may be illustrated for copper by fusion with XRF, where 99% of the time (1- $\alpha$ =0.99) at least 95% of subsamples ( $\rho$ =0.95) will have concentrations lying between 0.566 and 0.596 wt.%. Put more precisely, this means that if the same number of subsamples were taken and analysed in the same manner repeatedly, 99% of the tolerance intervals so constructed would cover at least 95% of the total population, and 1% of the tolerance intervals would cover less than 95% of the total population (ISO Guide 35). Please note that tolerance limits pertain to the homogeneity of the CRM only and should not be used as control limits for laboratory performance.

Based on the statistical analysis of the results of the inter-laboratory certification program it can be concluded that OREAS 553 is sufficiently homogenous and fit-for-purpose as a certified reference material (see 'Intended Use' below).

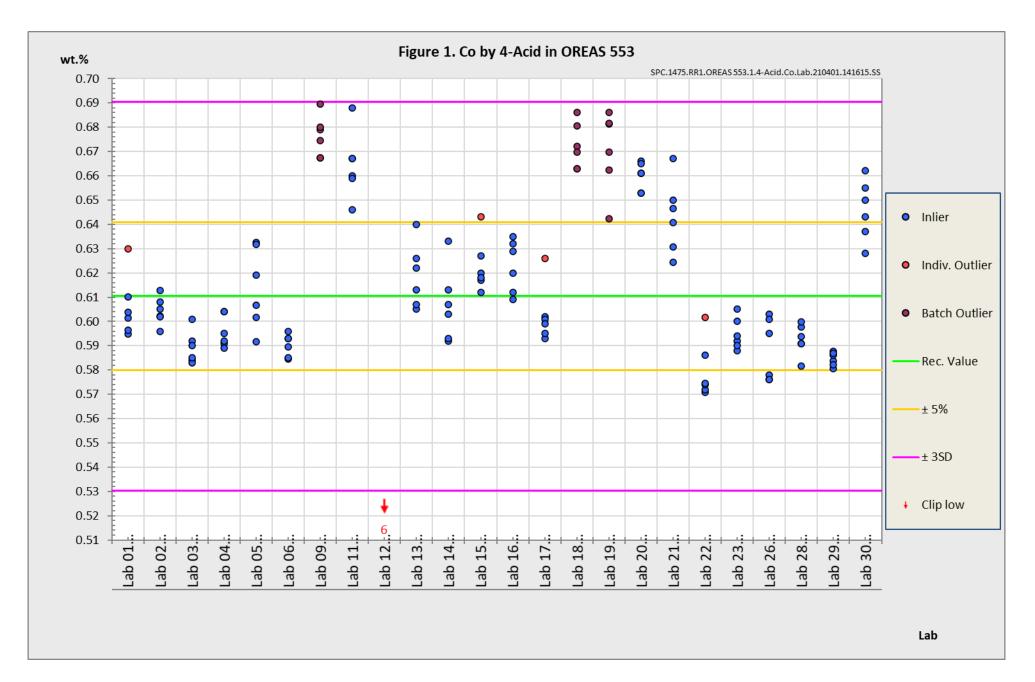


# PARTICIPATING LABORATORIES

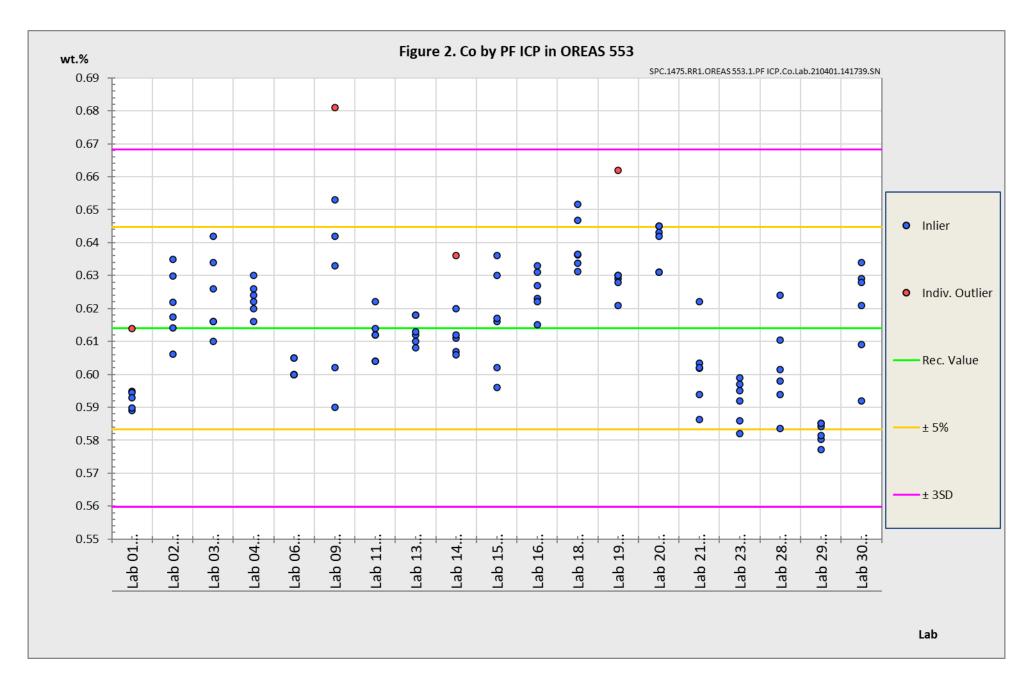
- 1. Actlabs, Ancaster, Ontario, Canada
- 2. AGAT Laboratories, Mississauga, Ontario, Canada
- 3. AH Knight, Kitwe, Copperbelt, Zambia
- 4. AH Knight, Lumwana, Mwinilunga, Zambia
- 5. ALS, Brisbane, QLD, Australia
- 6. ALS, Johannesburg, South Africa
- 7. ALS, Lima, Peru
- 8. ALS, Loughrea, Galway, Ireland
- 9. ALS, Vancouver, BC, Canada
- 10. American Assay Laboratories, Sparks, Nevada, USA
- 11. Bureau Veritas Commodities Canada Ltd, Vancouver, BC, Canada
- 12. Bureau Veritas Geoanalytical, Adelaide, SA, Australia
- 13. Bureau Veritas Geoanalytical, Perth, WA, Australia
- 14. Carsurin, Kendari, Sulawesi, Indonesia
- 15. Inspectorate (BV), Lima, Peru
- 16. Intertek Genalysis, Perth, WA, Australia
- 17. Intertek Testing Services Philippines, Cupang, Muntinlupa, Philippines
- 18. MSALABS, Vancouver, BC, Canada
- 19. Nagrom, Perth, WA, Australia
- 20. Ontario Geological Survey, Sudbury, Ontario, Canada
- 21. PT Geoservices Ltd, Cikarang, Jakarta Raya, Indonesia
- 22. PT Intertek Utama Services, Jakarta Timur, DKI Jakarta, Indonesia
- 23. Robinson International Afrique SAS, Lubumbashi, Katanga, Democratic Republic of the Congo
- 24. Ruashi Mine Site Laboratory, Lubumbashi, Katanga, Democratic Republic of the Congo
- 25. Saskatchewan Research Council, Saskatoon, Saskatchewan, Canada
- 26. SGS, Randfontein, Gauteng, South Africa
- 27. SGS Canada Inc., Vancouver, BC, Canada
- 28. SGS del Peru, Lima, Peru
- 29. SGS Geosol Laboratorios Ltda, Vespasiano, Minas Gerais, Brazil
- 30. UIS Analytical Services, Centurion, South Africa

Please note: To preserve anonymity, the above numbered alphabetical list of participating laboratories does not correspond with the Lab ID numbering on the scatter plots below.

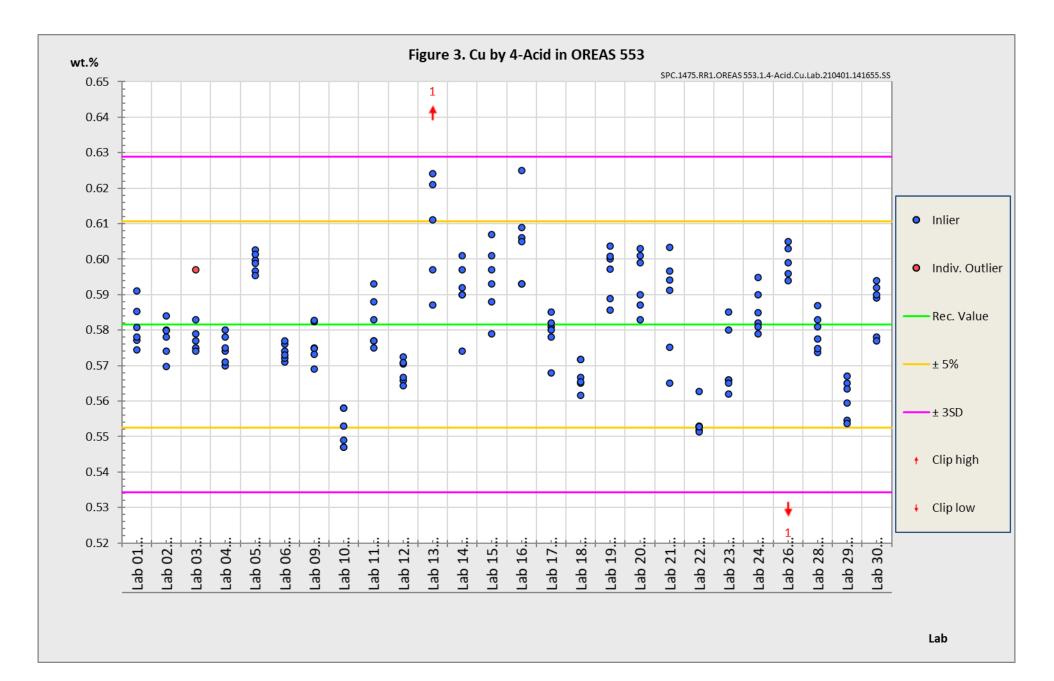




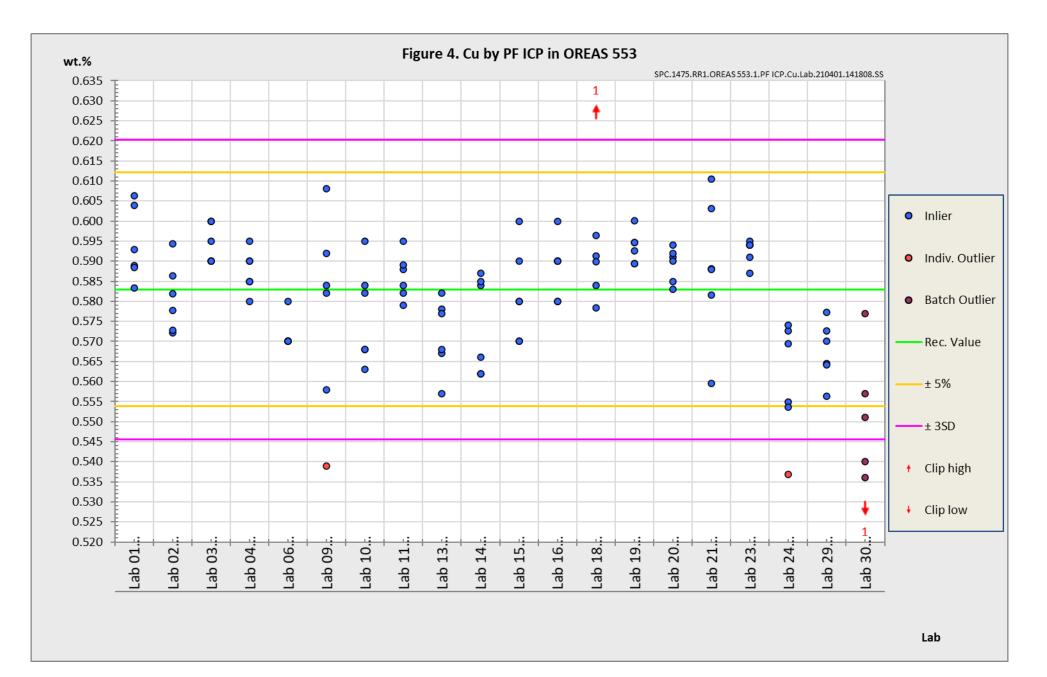














## PREPARER AND SUPPLIER

Certified reference material OREAS 553 was prepared and certified by:



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## METROLOGICAL TRACEABILITY

The analytical samples were selected in a manner to represent the entire batch of prepared CRM. This 'representivity' was maintained in each submitted laboratory sample batch and ensures the user that the data is traceable from sample selection through to the analytical results that underlie the consensus values. Each analytical data set has been validated by its assayer through the inclusion of internal reference materials and QC checks during analysis.

The laboratories were chosen on the basis of their competence (from past performance in inter-laboratory programs undertaken by ORE Pty Ltd) for a particular analytical method, analyte or analyte suite, and sample matrix. Most of these laboratories have and maintain ISO 17025 accreditation. The certified values presented in this report are calculated from the means of accepted data following robust statistical treatment as detailed in this report.

Guide ISO/TR 16476:2016, section 5.3.1 describes metrological traceability in reference materials as it pertains to the transformation of the measurand. In this section it states, *"Although the determination of the property value itself can be made traceable to appropriate units through, for example, calibration of the measurement equipment used, steps like the transformation of the sample from one physical (chemical) state to another cannot. Such transformations may only be compared with a reference (when available), or among themselves. For some transformations, reference methods have been defined and may be used in certification projects to evaluate the uncertainty associated with such a transformation. In other cases, only a comparison among different laboratories using the same method is possible. In this case, certification takes place on the basis of agreement among independent measurement results (see ISO Guide 35:2006, Clause 10)."* 

## COMMUTABILITY

The measurements of the results that underlie the certified values contained in this report were undertaken by methods involving pre-treatment (digestion/fusion) of the sample. This served to reduce the sample to a simple and well understood form permitting calibration using simple solutions of the CRM. Due to these methods being well understood and highly effective, commutability is not an issue for this CRM. Being matrix-matched, OREAS 553 will display similar behaviour in the relevant measurement process to the routine 'field' samples for which OREAS 553 is designated to monitor. To maintain commutability, care should be taken to always ensure 'matrix matching' as close as practically achievable. The matrix and mineralisation style of the CRM is described in the 'Source Material' section and users should select appropriate CRMs matching these attributes to their field samples.



## INTENDED USE

OREAS 553 is intended to cover all activities needed to produce a measurement result. This includes extraction, possible separation steps and the actual measurement process (the signal producing step). OREAS 553 may be used to calibrate the entire procedure by producing a pure substance CRM transformed into a calibration solution. OREAS 553 is intended for the following uses:

- For the monitoring of laboratory performance in the analysis of analytes reported in Table 1 in geological samples;
- For the verification of analytical methods for analytes reported in Table 1;
- For the calibration of instruments used in the determination of the concentration of analytes reported in Table 1.

## STABILITY AND STORAGE INSTRUCTIONS

OREAS 553 has been prepared from a blend of copper-cobalt oxide ore samples. It is low in reactive sulphide (0.28 wt.% S) and in its unopened state and under normal conditions of storage has a shelf life beyond ten years. Its stability will be monitored at regular intervals and purchasers notified if any changes are observed.

#### Single-use sachets

Following analysis of the CRM subsample it is the manufacturers' expectation that any remaining material is discarded. The stability of the material after opening the sachet is not within the scope of proper use. However, if opened sachets are resealed after opening, then under ordinary\* storage conditions the CRM will have a shelf-life beyond ten years.

\*ordinary storage conditions: means storage not in direct sunlight in a dry, clean, well ventilated area at temperatures between -5° and 50°C.

#### Repeat-use packaging (e.g., 1kg plastic jars)

The stability of the CRM after opening the lid of the plastic jar is only affected by local atmospheric conditions with regard to oxidation and hygroscopic change. There is no segregation affect (please see our <u>Technical Note on Particle Segregation</u>).

The primary cause of change through oxidation is in relation to the breakdown of sulphide minerals to sulphates and is negligible for OREAS 553 given its low sulphur concentration (0.28 wt.% S).

Hygroscopic change is the amount of absorbed moisture (weakly held H<sub>2</sub>O- molecules on the surface of exposed material) following exposure to the local atmosphere. Usually, equilibration of material to the local atmosphere will only occur if the material is spread into a thin (~2mm thick) layer and left exposed for a period of 2 hours. OREAS 553 contains a non-hygroscopic matrix and therefore, exposure to a local atmosphere that is significantly different (in terms of temperature and humidity) from the climate during manufacturing will have negligible impact on the precision of results. The 'Physical Properties' section indicates the approximate moisture concentration.



## INSTRUCTIONS FOR CORRECT USE

The certified values for OREAS 553 refer to the concentration level in its packaged state. It should not be dried prior to weighing and analysis.

1kg jars permit repeated sampling as long as the lid is promptly re-secured to prevent airborne contamination.

#### Minimum sample size

As a practical guide, the minimum mass of sample used should match the typical mass that the laboratories used in the interlaboratory (round robin) certification program. This means that different sample masses should be used depending on the operationally defined methodology.

- Lithium borate fusion with X-ray fluorescence finish: ≥0.2g;
- Loss on Ignition (LOI) at 1000°C: ≥1g;
- C and S by infrared combustion furnace/CS analyser: ≥0.1g;
- Cu and Co by 5% sulphuric acid leach with ICP or AAS finish: 0.5g;
- Cu and Co by 10% sulphuric acid leach with ICP or AAS finish: ≥0.5g;
- Sodium peroxide fusion with ICP-OES and/or MS finish: ≥0.2g;
- 4-acid digestion with ICP-OES and/or MS finish: ≥0.25g;
- Aqua regia digestion with ICP-OES and/or MS finish: ≥0.5g.

#### QC monitoring using multiples of the Standard Deviation (SD)

In the application of SD's in monitoring performance it is important to note that not all laboratories function at the same level of proficiency and that different methods in use at a particular laboratory have differing levels of precision. Each laboratory has its own inherent SD (for a specific concentration level and analyte-method pair) based on the analytical process and this SD is not directly related to the round robin program.

The majority of data generated in the round robin program was produced by a selection of world class laboratories. The SD's thus generated are more constrained than those that would be produced across a randomly selected group of laboratories. To produce more generally achievable SD's the 'pooled' SD's provided in this report include inter-laboratory bias. This 'one size fits all' approach may require revision at the discretion of the QC manager concerned following careful scrutiny of QC control charts.

The performance gates shown in Table 1 are intended only to be used as a first principle guide as to what a laboratory may be able to achieve. Over a period of time monitoring your own laboratory's data for this CRM, SD's should be calculated directly from your own laboratory's process. This will enable you to establish more specific performance gates that are fit for purpose for your application as well as the ability to monitor bias. If your long-term trend analysis shows an average value that is within the 95% confidence interval then generally there is no cause for concern in regard to bias.

It is important to note that in the analytical industry there is no standardisation of the aqua regia digestion process nor the 5% and 10% sulphuric acid leach process. These methods are partial empirical digests and differences in recoveries for various analytes are commonplace. These are caused by variations in the digest conditions and in the case for aqua regia digestion can include the ratio of nitric to hydrochloric acids, acid strength, temperatures, leach times and secondary digestions. Recoveries for sulphide-hosted base



metal sulphides approach total values, however, other analytes, in particular the lithophile elements, show greater sensitivity to method parameters. This can result in lack of consensus in an inter-laboratory certification program for these elements.

The approach applied here is to report certified values in those instances where reasonable agreement exists amongst a majority of participating laboratories. The results of specific laboratories may differ significantly from the certified values, but will, nonetheless, be valid and reproducible in the context of the specifics of the aqua regia method in use. Users of this reference material should, therefore, be mindful of this limitation when applying the certified values in a quality control program.

## HANDLING INSTRUCTIONS

Fine powders pose a risk to eyes and lungs and therefore standard precautions such as the use of safety glasses and dust masks are advised.

# LEGAL NOTICE

Ore Research & Exploration Pty Ltd has prepared and statistically evaluated the property values of this reference material to the best of its ability. The Purchaser by receipt hereof releases and indemnifies Ore Research & Exploration Pty Ltd from and against all liability and costs arising from the use of this material and information.

## DOCUMENT HISTORY

Revision No.	Date	Changes applied
1	21 <sup>st</sup> April, 2021	Applied minor statistical correction to Cobalt by XRF method results.
0	7 <sup>th</sup> April, 2021	First publication.

# QMS ACCREDITATION

ORE Pty Ltd is accredited to ISO 9001:2015 by Lloyd's Register Quality Assurance Ltd for its quality management system including development, manufacturing, certification and supply of CRMs.



# **CERTIFYING OFFICER**



21<sup>st</sup> April, 2021

Craig Hamlyn (B.Sc. Hons - Geology), Technical Manager - ORE P/L



## REFERENCES

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ISO 16269:2014, Statistical interpretation of data – Part 6: Determination of statistical tolerance intervals.

ISO/TR 16476:2016, Reference Materials – Establishing and expressing metrological traceability of quantity values assigned to reference materials.

ISO 17025:2005, General requirements for the competence of testing and calibration laboratories.

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## APPENDIX

For Cu and Co by the two sulphuric acid leaches, specific methodologies were detailed for the participating laboratories to follow:

#### 5% sulphuric acid leach

- 1. Weigh  $0.500 \pm 0.002g$  of sample pulp into a clean 250 ml flask.
- 2. Add to the flask 0.5g of Sodium Sulphite (AR Grade).
- 3. Add 50 ml of approximately 55 g/l Sulphuric acid solution (prepared from a 98% pure concentrated sulphuric acid).
- 4. Put the cap on the flask and start automatic shaking. Leave the sample on continuous shaking to leach for four (4) hours.
- 5. Remove the cap and add 25 ml of hydrochloric acid.
- 6. Dilute to a final volume of 250 ml with distilled/deionised water up to the mark and mix again by inverting at least 10 times. Allow the solution to settle for 30 minutes.
- 7. The solution is now ready to be analysed by ICP or AAS.

#### 10% sulphuric acid leach

- 1. Weigh  $0.500 \pm 0.002g$  of sample pulp into a clean 250 ml flask.
- 2. Add to the flask 0.5g of Sodium Sulphite (AR Grade).
- 3. Add 50 ml of approximately 110 g/l Sulphuric acid solution (prepared from a 98% pure concentrated sulphuric acid).
- 4. Put the cap on the flask and start automatic shaking. Leave the sample on continuous shaking to leach for four (4) hours.
- 5. Remove the cap and add 25 ml of hydrochloric acid.
- 6. Dilute to a final volume of 250 ml with distilled/deionised water up to the mark and mix again by inverting at least 10 times. Allow the solution to settle for 30 minutes.
- 7. The solution is now ready to be analysed by ICP or AAS.

